BODNEVAS, A., red.; VISHOMIRSKIS, R.[Visomirskis, R.], red.;
GAL'DIKENE, O.[Galdikiene, O.], red.; MATULIS, Yu.
[Matulis, J.], red.; PETRAUSKAS, V., red.; KARVYALIS, V.
[Karvelis, V.], tekhn. red.

[Theory and practice of bright electroplating] Teoriia i praktika blestiashchikh gal'vanopokrytii; osnovnye materialy. Vilnius, Gos.izd-vo polit. i nauchn. lit-ry Litovskoi SSR, 1963. 366 p. (MIRA 17:1)

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IEBEDNIKAS, B.I. [Lebednykas, B.]; MOLCHADSKIY, A.M. [Molcadakis, A.]; MATULIS, Yu.Yu. [Matulis, J.]; VISHOMIPSKIS, R.M. [Visomirskis, R.]

Influence of some factors on cathode processes during the electrodeposition of silver from cyanide electrolytes. Trudy AN Lit. SSR. Ser. B. no.2:13-24 '65. (MIRA 19:2)

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VISHOMIPSKIS, R.M. [Visomirskis, k.]; SHIVITSKIS, Yu.P. [Sivickis, J.]

Stability of electrochemical systems during the separation of metals and hydrogen. Trudy AN Lit.SSR. Ser. B. no.2:25-35 '65.

(MIRA 19:2)

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BORODULINA, V.V.; BOLDYREVA, G.G.; VISHOMIRSKIS, R.M. [Visomirskis, R.]; MOLCHADSKIY, A.M.

Study of the process of electrodeposition of palladium from tetrammine chlorite solutions. Trudy AN Lit.SSR. Ser. B. no.2: 49-59 165. (MIRA 19:2)

1. Institut khimil i khimicheskoy tekhnologii AN Litovskoy SSR. Submitted September 22, 1964.

PILITE, S.P. [Pilyte, S.]; VISHOMIRSKIS, R.M. [Visomirskis, R.];
MOLCHADSKIY, A.M.

Electrodeposition of palladium from ethylenediamine electrolytes.
Trudy AN Lit.SSR. Ser. B. no.2:61-72 '65. (MIRA 19:2)

1. Institut khimii i khimicheskoy tekhnologii AN Litovskoy SSR.
Submitted September 26, 1964.

VISHOMIRSKIS, R.M. [Visomirskis, R.]: STYAPONAVICHYUS, A.A. [Steponavicius, A.]

Concentration changes in a cathode layer during the electrodeposition of copper from cyanide electrolytes. Trudy AN Lit.SSR. Ser. B no.3:49-62 165. (MIRA 19:1)

1. Institut khimii i khimicheskoy takhnologii AM Litavskoy SSR.

LJP(c) EWT(m)/EWP(t) L 17992-66

SOURCE CODE: UR/0236/65/000/004/0015/0026 ACC NR: AP6004533

AUTHOR: Pilite, S. P. (Pilyte, S.); Morgenshtern, Ya. L. (Morgensternas, J.); Molchadskiy, A. M. (Molcadskis, A.); Vishomirskis, R. M. (Visomirskis, R.)

ORG: Institute of Chemistry and Chemical Engineering, Academy of Sciences Lithua nian SSR (Institut khimii i khimicheskoy tekhnologii Akademii nauk Litovskoy SSR)

TITLE: Study of electrodeposition of palladium from phosphate solutions 65,27 44.55 16

SOURCE: AN LitSSR. Trudy. Seriya B. Fiziko-matematicheskiye, khimicheskiye, geologicheskiye i tekhnicheskiye nauki, no. 4, 1965, 15-26

TOPIC TAGS: cathode polarization, palladium, electrodeposition

ABSTRACT: In order to elucidate the processes taking place during electrodeposition of palladium from phosphate solutions, the cathodic polarization of this metal was measured as a function of the composition, temperature, and rate of stirring of the electrolyte. The following electrolyte compositions were employed (in g/1): (a) Pd (as PdCl₂), 4.8; Na₂HPO₄·12H₂O, 100; (NH₄)₂HPO₄, 20; NH₄Cl, 25; (b) Pd (as (PdCl2), 4.8; Na2HPO4.12H20, 100; (NH4)2HPO4, 20; C6H5COOH, 2.5. It was found that the cathodic polarization curves consist of four main branches. The metal deposi-

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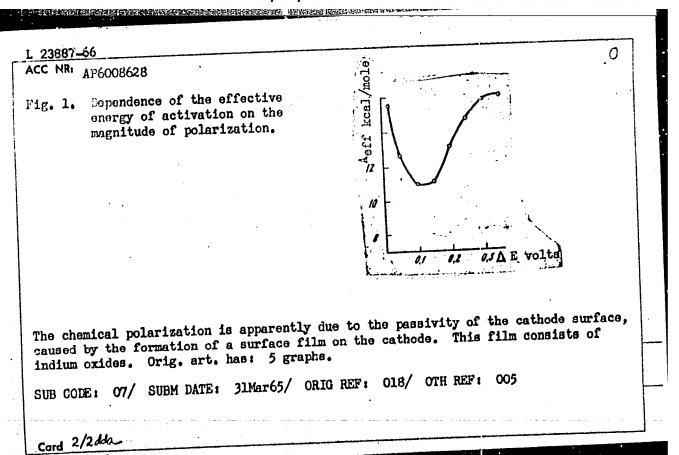
L 17992-66 ACC NR: AP6004533

tion takes place on the second branch, which is separated from the steady state potential of the electrode by more than 0.4 v. Replacement of ammonium chloride of benzoic acid causes the cathodic polarization of palladium to increase. Using a temperature-kinetic method and a rotating disc cathode, the authors showed that the electrodeposition is associated with concentration and chemical polarization. The nature of polarization changes with the potential and solution composition. On the basis of the results, it is postulated that the substantial cathodic polarization in phosphate electrolytes is due to the passivation of the electrode surface during electrolysis. Orig. art. has: 8 figures, 1 table.

SUB CODE: 07/ SUBM DATE: 04May65/ ORIG REF: 012/ OTH REF: 004

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L 23887-66 EWI(m)/EMP(t) IJP(c) JD ACC NRI AP6008628 SOURCE CODE: UR/0365/65/001/006/0703/0708
AUTHORS: Voronko, A. A. (deceased); Vishomirskis, R. M.; Molchadskiy, A. M.
ORG: Academy of Science, Lithuanian SSR, Institute for Chemistry and Chemical
Technology (Akademiya mauk Litovskoy SSR, Institut khimii i khimicheskoy tekhnologii)
TITLE: The kinetics of electrodeposition of indium from sulfate solutions
SOURCE: Zashchita metallov, v. 1, no. 6, 1965, 703-708
TOPIC TAGS: electrodeposition, electroplating, indium, indium sulfate, cathode
polarization ABSTRACT: This investigation was conducted to resolve existing differences concern-
deposition of indium was determined. A copper strip of 1-cm ² area covered with a 5-micron thick indium layer served as the cathode. One side of the cathode was insulated with PKhV-26 lacquer. The anode consisted of metallic indium (area of
2) Who amendmental regults are presented graphically (see Fig. 1). It was -
found that the polarization curves exhibited a minimum in the regions of residue of the dependence on the concentration and pil of
11 - 1-11- TI in shorm that the ghorn decrease in the rate of the cavillate
process is due to chemical changes which give rise to concentration polarization.
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INVENTOR: Shivitskis, Yu. P.; Vesene, T. B.; Vishomirskis, R. H. ORG: none TITLE: Method of electroplating with cadmium. Class 48, No. 180933 [announced by the Institute of Chemistry and Chemical Technology AN Lithuanian SSR (Institut khimii i kimicheskoy tekhnologii AN Litovskoy SSR) SOURCE: Izobreteniya, promyshlennyye obraztsy, tovarnyye znaki, no. 8, 1966, 123 TOPIC TAGS: cadmium, cadmium plating, cadmium electroplating ABSTRACT: This Author Certificate introduces a method of electroplating with cadmium in an electrolyte containing 30 g/l cadmium oxide, 106 g/l potassium cyanide cadmium in an electrolyte containing 30 g/l cadmium in a lustrous coating 10 g/l caustic potash, and 14 g/l potassium carbonate. To obtain a lustrous coating non-naphthylamine sulfate is added to the initial solution to saturation and the groupses wetting agent is added in the amount of 2.5 ml/l. Electrolysis is carrivereses wetting agent is added in the amount of 2.5 ml/l. Electrolysis is carriveresed current with a cathodic period of 15 sec and an anodic period of 3 sec. SUB CODE: 13/ SUBM DATE: 06Mar64/ ATD PRESS: 4 2 45 Cord 1/1 Mar.	 L 21196-66 EWT(m)/EWP(t) IJP(c) JD SOURCE CODE: UR/0413/66/000,'008/0123/0123
TITLE: Method of electroplating with cadmium. Class 48, No. 180933 [announced by the Institute of Chemistry and Chemical Technology AN Lithuanian SSR (Institut khimii i kimicheskoy tekhnologii AN Litovskoy SSR) SOURCE: Izobreteniya, promyshlennyye obraztsy, tovarnyye znaki, no. 8, 1966, 123 TOPIC TAGS: cadmium, cadmium plating, cadmium electroplating ABSTRACT: This Author Certificate introduces a method of electroplating with cadmium in an electrolyte containing 30 g/l cadmium oxide, 106 g/l potassium cyanide 10 g/l caustic potash, and 14 g/l potassium carbonate. To obtain a lustrous coating 10 g/l caustic potash, and 14 g/l potassium carbonate. To obtain a lustrous coating 10 manipulation and the canaphthylamine sulfate is added to the initial solution to saturation and the "Progress" wetting agent is added in the amount of 2.5 ml/l. Electrolysis is cartical out at 18—20C at a current density of 1.3—3.5 a/dm with a periodically ried out at 18—20C at a current density of 1.3—3.5 a/dm with a periodically ried out at 18—20C at a cathodic period of 15 sec and an anodic period of 3 sec. [ND] SUB CODE: 13/ SUBM DATE: 06Mar64/ ATD PRESS: 4245	ALC INI MI UULUUL
TITLE: Method of electroplating with cadmium. Class 48, No. 180933 [announced by the Institute of Chemistry and Chemical Technology AN Lithuanian SSR (Institut khimil i kimicheskoy tekhnologii AN Litovskoy SSR) SOURCE: Izobreteniya, promyshlennyye obraztsy, tovarnyye znaki, no. 8, 1966, 123 TOPIC TAGS: cadmium, cadmium plating, cadmium electroplating ABSTRACT: This Author Certificate introduces a method of electroplating with cadmium in an electrolyte containing 30 g/1 cadmium oxide, 106 g/1 potassium cyanide 10 g/1 caustic potash, and 14 g/1 potassium carbonate. To obtain a lustrous coating 10 g/1 caustic potash, and 14 g/1 potassium carbonate. To obtain a lustrous coating 10 g/1 caustic potash, and 14 g/1 potassium carbonate. To obtain a lustrous coating 10 g/1 caustic potash, and 14 g/1 potassium carbonate. To obtain a lustrous coating 10 g/1 caustic potash, and 14 g/1 potassium carbonate. To obtain a lustrous coating 10 g/1 caustic potash, and 14 g/1 potassium carbonate. To obtain a lustrous coating 10 g/1 caustic potash, and 14 g/1 potassium oxide, 106 g/1 potassium cyanide cadmium in an electrolyte containing 30 g/1 cadmium oxide, 106 g/1 potassium cyanide cadmium in an electrolyte containing 30 g/1 cadmium oxide, 106 g/1 potassium cyanide cadmium in an electrolyte containing 30 g/1 cadmium oxide, 106 g/1 potassium cyanide cadmium in an electrolyte containing 30 g/1 cadmium oxide, 106 g/1 potassium cyanide cadmium in an electrolyte containing 30 g/1 cadmium oxide, 106 g/1 potassium cyanide cadmium in an electrolyte containing 30 g/1 cadmium oxide, 106 g/1 potassium cyanide cadmium in an electrolyte containing 30 g/1 cadmium oxide, 106 g/1 potassium cyanide cadmium in an electrolyte containing 30 g/1 cadmium oxide, 106 g/1 potassium cyanide cadmium in an electrolyte containing 30 g/1 cadmium oxide, 106 g/1 potassium cyanide cadmium cyanide cadmium in an electrolyte containing 30 g/1 cadmium oxide, 106 g/1 potassium cyanide cadmium cyanide cadmium cyanide cadmium cyanide cadmium cyanide cadmium cyanide c	INVENTOR: Shivitskis, Yu. P.; Vesene, T. B.; Vishomirskis, K. T.
khimil i kimicheskoy tekhnologii AN Litovskoy SSR) SOURCE: Izobreteniya, promyshlennyye obraztsy, tovarnyye znaki, no. 8, 1966, 123 TOPIC TAGS: cadmium, cadmium plating, cadmium electroplating ABSTRACT: This Author Certificate introduces a method of electroplating with cadmium in an electrolyte containing 30 g/l cadmium oxide, 106 g/l potassium cyanide, cadmium in an electrolyte containing 30 g/l cadmium oxide, 106 g/l potassium cyanide, cadmium in an electrolyte containing 30 g/l cadmium oxide, 106 g/l potassium cyanide, cadmium in an electrolyte containing 30 g/l cadmium oxide, 106 g/l potassium cyanide, cadmium in an electrolyte containing 30 g/l cadmium oxide, 106 g/l potassium cyanide, cadmium in an electrolyte containing 30 g/l cadmium oxide, 106 g/l potassium cyanide, cadmium in an electrolyte containing 30 g/l cadmium oxide, 106 g/l potassium cyanide, cadmium in an electrolyte containing 30 g/l cadmium oxide, 106 g/l potassium cyanide, cadmium in an electrolyte containing 30 g/l cadmium oxide, 106 g/l potassium cyanide, cadmium oxide, 106 g/l potassium	/ b A
khimif i kimicheskoy teknilosoga solution solution (no. 8, 1966, 123) SOURCE: Izobreteniya, promyshlennyye obraztsy, tovarnyye znaki, no. 8, 1966, 123 TOPIC TAGS: cadmium, cadmium plating, cadmium electroplating ABSTRACT: This Author Certificate introduces a method of electroplating with cadmium in an electrolyte containing 30 g/l cadmium oxide, 106 g/l potassium cyanide, 10 g/l caustic potash, and 14 g/l potassium carbonate. To obtain a lustrous coating 10 g/l caustic potash, and 14 g/l potassium carbonate. To obtain a lustrous coating 2-naphthylamine sulfate is added to the initial solution to saturation and the canaphthylamine sulfate is added in the amount of 2.5 ml/l. Electrolysis is care. Progress' wetting agent is added in the amount of 2.5 ml/l. Electrolysis is cared out at 18—20C at a current density of 1.3—3.5 a/dm³ with a periodically ried out at 18—20C at a current density of 1.3—3.5 a/dm³ with a periodically reversed current with a cathodic period of 15 sec and an anodic period of 3 sec. [ND] SUB CODE: 13/ SUBM DATE: 06Mar64/ ATD PRESS: 42.45	TITLE: Method of electropiating with the state of Chemistry and Chemical Technology AN Lithuanian SSR (Institute of Chemistry and Chemical Technology AN Lithuanian SSR (Institute of Chemistry and Chemical Technology AN Lithuanian SSR (Institute of Chemistry and Chemical Technology AN Lithuanian SSR (Institute of Chemistry and Chemical Technology AN Lithuanian SSR (Institute of Chemistry and Chemical Technology AN Lithuanian SSR (Institute of Chemistry and Chemical Technology AN Lithuanian SSR (Institute of Chemistry and Chemical Technology AN Lithuanian SSR (Institute of Chemistry and Chemical Technology AN Lithuanian SSR (Institute of Chemistry and Chemical Technology AN Lithuanian SSR (Institute of Chemistry and Chemical Technology AN Lithuanian SSR (Institute of Chemistry and Chemical Technology AN Lithuanian SSR (Institute of Chemistry and Chemical Technology AN Lithuanian SSR (Institute of Chemistry and Chem
SOURCE: Izobreteniya, promyshlennyye obraztsy, tovarnyye znaki, no. 8, 1906, 125 TOPIC TAGS: cadmium, cadmium plating, cadmium electroplating ABSTRACT: This Author Certificate introduces a method of electroplating with cadmium in an electrolyte containing 30 g/l cadmium oxide, 106 g/l potassium cyanide cadmium in an electrolyte containing 30 g/l cadmium oxide, 106 g/l potassium carbonate. To obtain a lustrous coating 10 g/l caustic potash, and 14 g/l potassium carbonate. To obtain a lustrous coating a-naphthylamine sulfate is added to the initial solution to saturation and the canaphthylamine sulfate is added in the amount of 2.5 ml/l. Electrolysis is car-"Progress" wetting agent is added in the amount of 2.5 ml/l. Electrolysis is car-"red out at 18—20C at a current density of 1.3—3.5 a/dm ³ with a periodically ried out at 18—20C at a current density of 1.3—3.5 a/dm ³ with a periodically reversed current with a cathodic period of 15 sec and an anodic period of 3 sec. [ND] SUB CODE: 13/ SUBM DATE: 06Mar64/ ATD PRESS: 4 2.45	
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ABSTRACT: This Author Certificate introduces a method of electropiating wanide, cadmium in an electrolyte containing 30 g/l cadmium oxide, 106 g/l potassium cyanide, 10 g/l caustic potash, and 14 g/l potassium carbonate. To obtain a lustrous coating, 10 g/l caustic potash, and 14 g/l potassium carbonate. To obtain a lustrous coating, 10 g/l caustic potash, and 14 g/l potassium carbonate. To obtain a lustrous coating, 10 g/l caustic potash, and 14 g/l potassium carbonate. To obtain a lustrous coating, 10 g/l caustic potash, and 14 g/l potassium cyanide, carbonate. To obtain a lustrous coating, 10 g/l caustic potash, and 14 g/l potassium cyanide, carbonate, 10 g/l potassium cyanide, cadmium oxide, 106 g/l potassium carbonate, To obtain a lustrous cadmium oxide, 106 g/l potassium carbonate, 107 g/l potassium carbonate, 107 g/l potassium cyanide, 107 g/l potassium carbonate, 107 g/l potassium c	SOURCE. Izobreteniya, promyshlennyyo
ABSTRACT: This Author Certificate introduces a method of electropiating wanide, cadmium in an electrolyte containing 30 g/l cadmium oxide, 106 g/l potassium cyanide, 10 g/l caustic potash, and 14 g/l potassium carbonate. To obtain a lustrous coating, 10 g/l caustic potash, and 14 g/l potassium carbonate. To obtain a lustrous coating, 10 g/l caustic potash, and 14 g/l potassium carbonate. To obtain a lustrous coating, 10 g/l caustic potash, and 14 g/l potassium carbonate. To obtain a lustrous coating, 10 g/l caustic potash, and 14 g/l potassium carbonate. To obtain a lustrous coating, 10 g/l caustic potash, and 14 g/l potassium carbonate. To obtain a lustrous coating, 10 g/l caustic potash, and 14 g/l potassium carbonate. To obtain a lustrous coating, 10 g/l caustic potash, and 14 g/l potassium carbonate. To obtain a lustrous coating, 10 g/l caustic potash, and 14 g/l potassium carbonate. To obtain a lustrous coating, 10 g/l caustic potash, and 14 g/l potassium carbonate. To obtain a lustrous coating, 10 g/l caustic potash, and 14 g/l potassium carbonate. To obtain a lustrous coating, 10 g/l caustic potash, and 14 g/l potassium carbonate. To obtain a lustrous coating, 10 g/l caustic potash, and 14 g/l potassium carbonate. To obtain a lustrous coating, 10 g/l caustic potash, and 14 g/l potassium carbonate. To obtain a lustrous coating, 10 g/l caustic potash, and 14 g/l potassium carbonate. To obtain a lustrous cathing the carbonate potash, and 14 g/l potassium carbonate. To obtain a lustrous cathing the carbonate potash, and 14 g/l potash, and 14	TOPIC TAGS: cadmium, cadmium plating, cadmium electropiating
"Progress" wetting agent is added in the amount of 2.5 ml/1. Electrically "Progress" wetting agent is added in the amount of 2.5 ml/1. Electrically ried out at 18—20C at a current density of 1.3—3.5 a/dm ³ with a periodically ried out at 18—20C at a current density of 1.5 sec and an anodic period of 3 sec. reversed current with a cathodic period of 15 sec and an anodic period of 3 sec. [ND] SUB CODE: 13/ SUBM DATE: 06Mar64/ ATD PRESS: 42.45	ABSTRACT: This Author Certificate introduces a method of electropiating washing cyanide, cadmium in an electrolyte containing 30 g/l cadmium oxide, 106 g/l potassium cyanide, cadmium in an electrolyte containing 30 g/l cadmium oxide, 106 g/l potassium carbonate. To obtain a lustrous coating, cadmium in an electrolyte containing and 14 g/l potassium carbonate.
SUB CODE: 13/ SUBM DATE: 06Mar64/ ATD PRESS: 4245	a-naphthylamine surface and is added in the amount of 2.5 ml. leading a periodically
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VISHOMIRSKIS, R.M. [Visomirskis, R.]; SHIVITSKIS, Yu.P. [Sivickis, J.]

THE THE PROPERTY OF THE PROPER

Self-oscillations in the electrodeposition of metals. Elektrokhimia 1 no.7:864-868 Jl *65. (MIRA 18:10)

1. Institut khimii i khimicheskoy tekhnologii AN Litovskoy SSR.

,	Investigating cathodic processes in an alkali palledium platting electrolyte. Zagnch, pro. 2 no.5:505-510 S-0 165. (MIRA) 18:9)
. • •	1. Institut khimij - khimicheskoy tekhnologii AN Litovskoy SSR.
	<i>,</i>

VORONKO, A.A. [deceased]; VISHOMIRSKIS, R.M.; MOLCHADSKIY, A.M.

Kinetics of the electrodeposition of indium from sulfate solutions.

Zashch.met. 1 no.6:703-708 N-D *65. (MIRA 12:11)

1. Institut khimii 1 khimicheskoy tekhnologii AN Litovskoy

SSR.

VESENE, T.B. [Vosiene, T.]; VISHOMIRSKIS, R.M. [Visomirskis, R.]

THE REPORT OF THE PROPERTY OF

Quantitative determination of the emulsifying agent "Progress" in nickel and cadmium plating baths. Trudy AN Lit. SSR. Ser.B no.1:3-6 '65. (MIRA 18:7)

1. Institut khimii 1 khimicheskoy tekhnologii AN Litovskoy SSR.

VISHRYAKOVA, T. P.

"Polymerisation of oleic acid and its methyl ester in presence of molecular compounds of boron fluoride and phosphonic acids." A. V. Topchiev and T. P. Vishryakova (p. 1618)

SO: Journal of General Chemistry (Zhurnal Obshchei Khimil) 1951, Vol 21, No 9.

WISHTAK, S.

Beets and Beet Sugar

Kolkh, proizy. No. 3,

For 800 centners of sugar beets per hectare. Kolkh. proizv. No. 3, 1953.

9. Monthly List of Russian Accessions, Library of Congress, June 1953, Uncl.

- VISHTAK, S.; KOKOSHKA, V. 1.
- USSR (600) 2.
- Woman as Farmers 4.
- Towards the sunny heights of communism, Hol. kolkh., 20, No. 1, 7. 1953.

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- 1. S. VISHTAK, V. KOKOSHKA
- 2. USSR (600)
- 4. Automobile Industry
- 7. Towards the sunny heights of communism. Mol, kolkh. 20 no. 1. 1953.

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VISHTAK, STEPANIDA

Women As Farmers

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VISHTAK, S.D. [Vyshtak, S.D.] svenevaya dvashdy Geroy Sotialisticheskogo Truda

Endorsed by experience. Nauka i shyttia 11 no.7:19-40 FK '61.

(MIRA 14:8)

1. Kolkhoz "Radyans'ka Ukraina" Grebenkovskogo rayona Kiyevskoy oblasti.

(Grebenka District—Sugar beets)

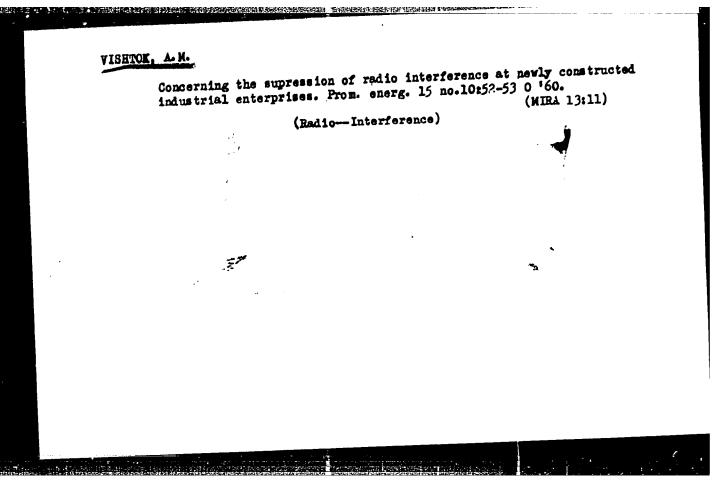
VISHTOK, A.M., inshener.

Suppressing radie noise in newly built industrial plants.

From.energ. 12 ne.9:29-33 S '57. (MIRA 10:10)

1.Moskovskiy preisvodstvenne-ekspluatatsionnyy etdel Gosudarstvennoge proektnoge instituts.

(Radio--Neise) (Metallurgical plants)



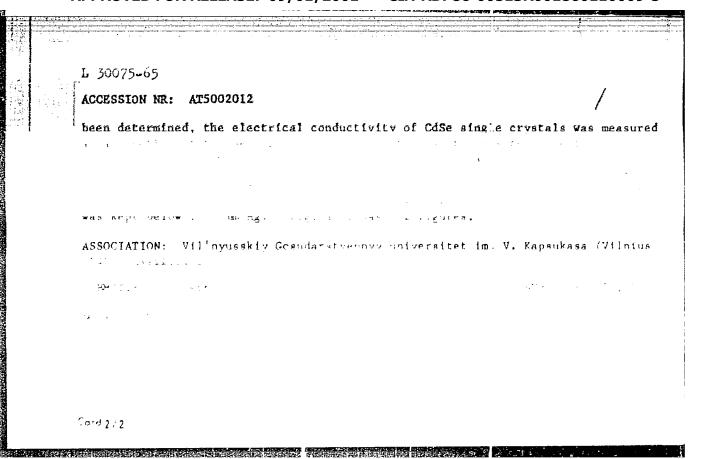
KOFMAN, K.D.; VISHTOK, V.M.; PANOVA, V.L., red.; IGLITSYN, I.L., red. izd-va; VORONIN, K.P., tekhn. red.

The state of the s

[Safety engineering and industrial hygiene regulations for electrical installation operations, in accordance with the May 8, 1960 decision, of the Central Committee of the Trade Union of Construction and Building Material Manufacturing Workers] Sbornik pravil tekhniki bezopasnosti i proizvodstvennoi sanitarii pri elektromontazhnykh rabotakh. Soglasovan s TsK profsoiuza rabochikh stroitel'stva i promyshlernosti stroitel'nykh materialov 7 maia 1960 g. Moskva, Gos.energ.izdvo, 1961. 255 p. (MIRA 15:2)

1. Russia (1917- T.S.F.S.R.) Glavnoye upravleniye po proizvodstvu elektromontazhnykh rabot. (Electric engineering--Safety measures)

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í	AUTHOR: Stonkus, S. I. (Stonkus, S.).	Vishvhakas, Yu. K. (Viscakas, J.)
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į	TITLE: The effect of the partial press	ure of selenium on the electrical conducti-
	vity of cadmium selenide single crystal	8
	SOURCE: AN LIESSR. Litovskiy fizichesk	ty sbornik, v. 4, 110. 2, 1964, 263-266
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VISI, I.

Technicians in the work contest; a letter to the editors. p. 5.

No. 2, Jan. 1955. MUSZAKI ELET Budapest

SOURCE: Monthly list of East European Accession, (FEAL), LC, VOL. 5, No. 3, March, 1956

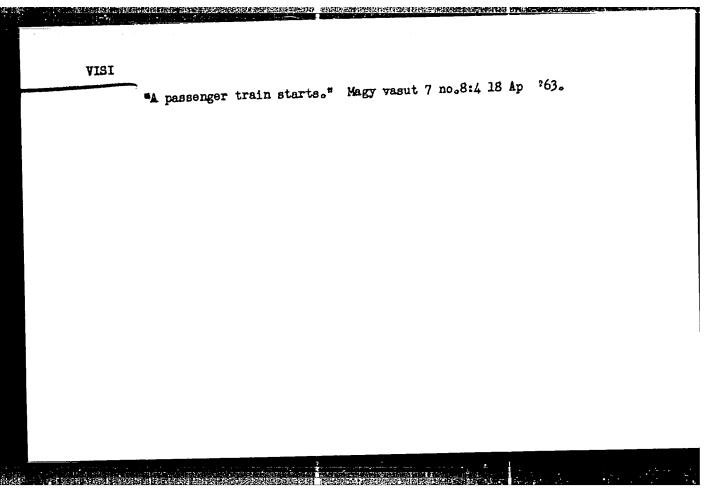
VASILEVSKIY, I.M.; VISHYAKOV, V.V.

Elastic scattering of 300 Mev. π -mesons on hydrogen. Zhur.eksp.i teor.fiz. 36 no.2:441-444 F 160. (MIRA 14:5)

1. O"yedinennyy institut yadernykh issledovaniy.
(Mesons—Scattering)

VISI

Fulfilling transportation tasks through better cooperation. Magy vasut 8 no.7:1,3 3 Ap 164.



VISI, Ferenc

Where forces are well concentrated. Magy vasut 8 no. 9: 3 4 My 164.

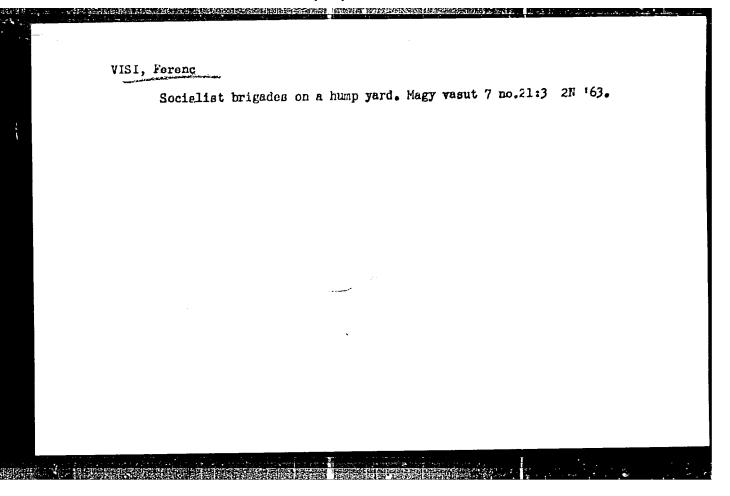
The Komarom railroadmen are well-prepared. Ibid.:4.

VISI, Ferenc

There is an increase in modern sleeping cars as well as eight-wheeled first-class cars in the rolling stock of the Hungarian State Railways. Vasut 13 no.1:8 30 Ja 163.

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Rostock Warmenunde; n itinerary. III. Magy vasut 7 no.6:5 18 Mr '63.



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Re 2N	cord traffic in Kiskunhalas too. Magy vasut 7 no.21:3

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Positive achievements by the Hungarian State Railways in the first half of the year. Magy vasut 8 no.14:1 18 J1 '64.

Trial run of the newest product of the Hungarian industry. Ibid.:1

Graduation day of railroad officers at the Szonyi Street Stadium. Ibid.:3

VISI, Ferenc

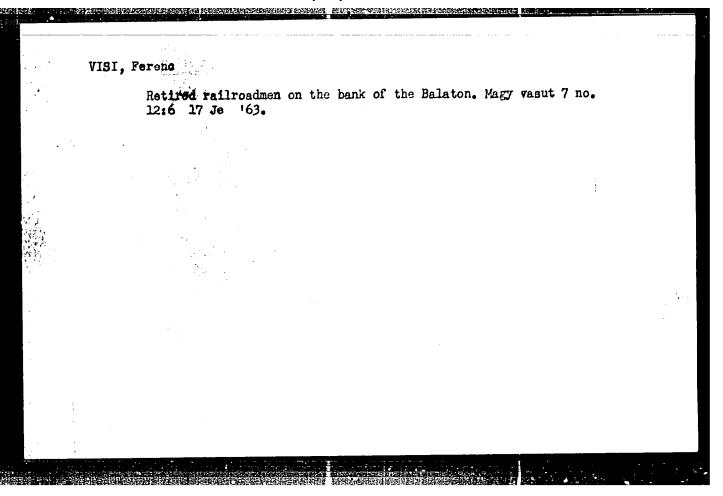
Density of passenger traffic maps are made for larger city districts. Magy vasut 8 no.15:1 1 Ag '64.

The international brigade has constructed a railroad line. Ibid.:1

What happened to them? Ibid .: 2

Hungarian State Railways finished last in the railroad competition among friendly nations. Ibid.:3

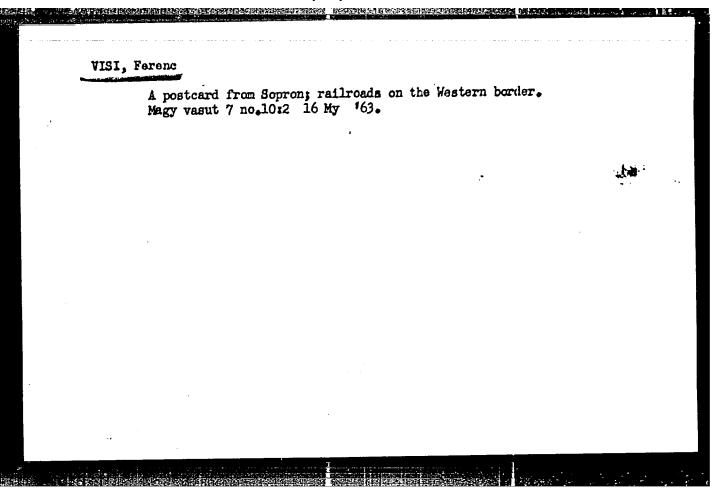
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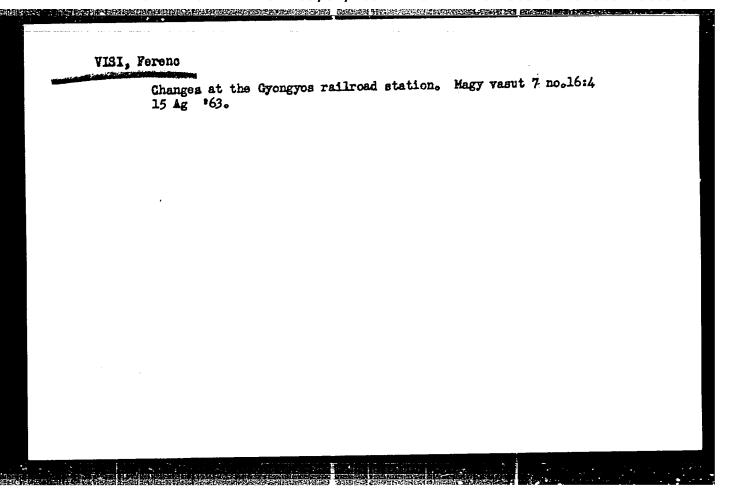


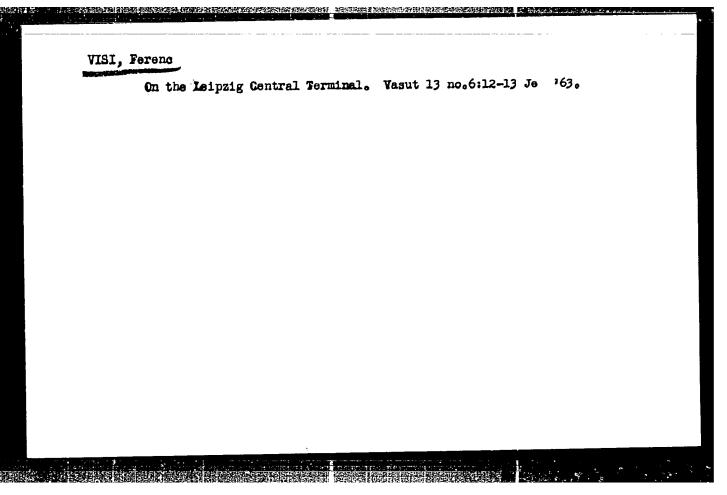
LINDNER, Jozsef; VISI, Ferenc

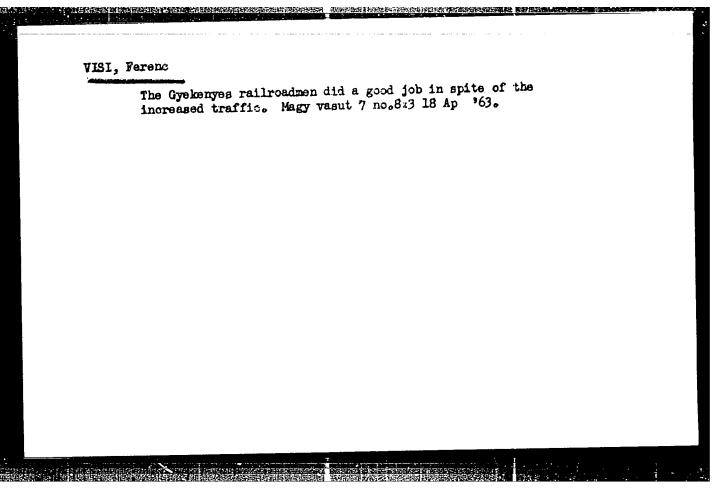
Reducing the turnaround time of railroad cars. Magy vasut 7 no.24: 3 14 D'63.

1. MAV vezerigazgatohelyettes. (for Lindner).





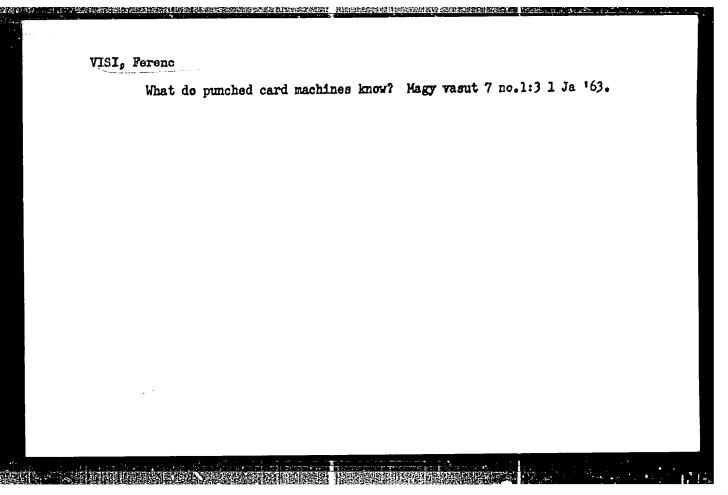


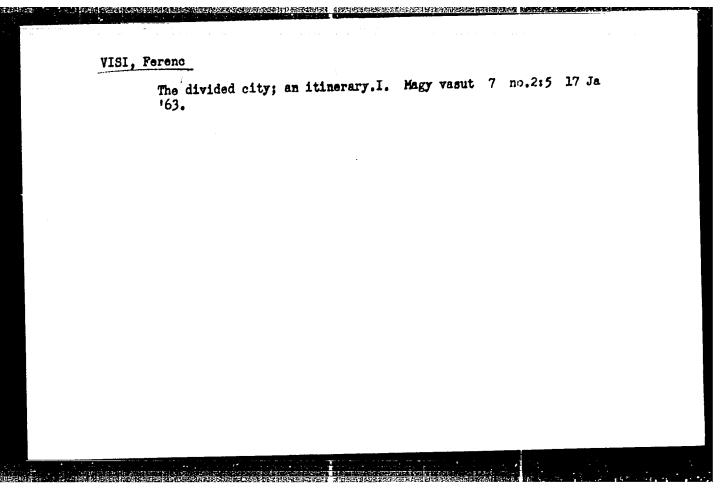


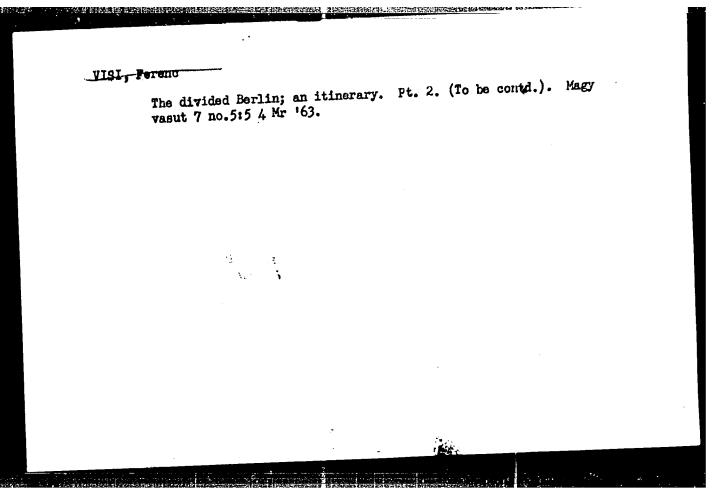
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What did the railroadmen of Rakosrendezo achieve in 1961? Vasut 12 no.2:19-20 25 F *62.







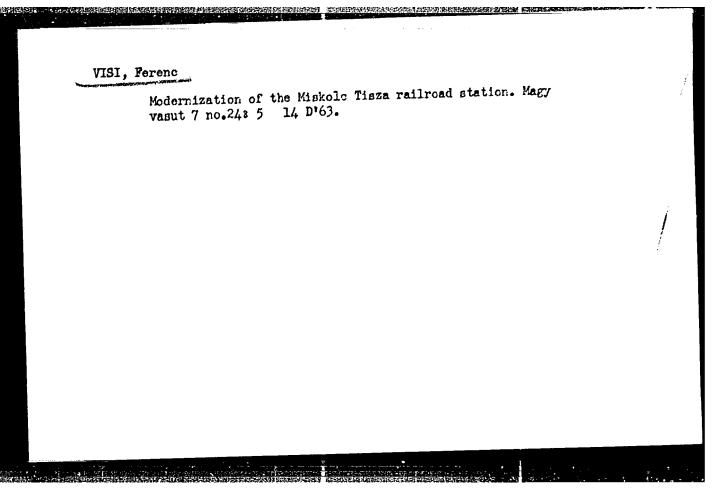
VISI, F	Morning tri	p on a slow	train.	Magy vasut 7 no.7:3 2 Ap 163.				
						£		
						,		

,	In spite of good cooperation there are many problems to be solved in Sturovo. Magy vasut 8 no.1:4 1 Ja 64	

VISI, Ferenc

Is it possible on one and impossible on other places?

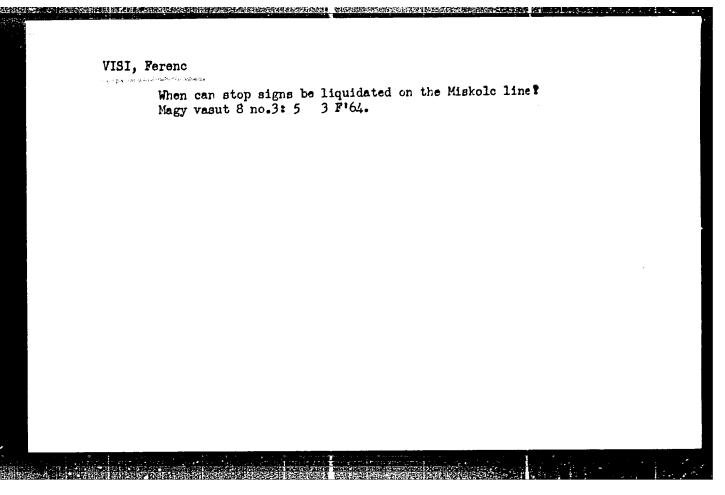
Magy vasut 7 no.13:4 2 Jl '63.

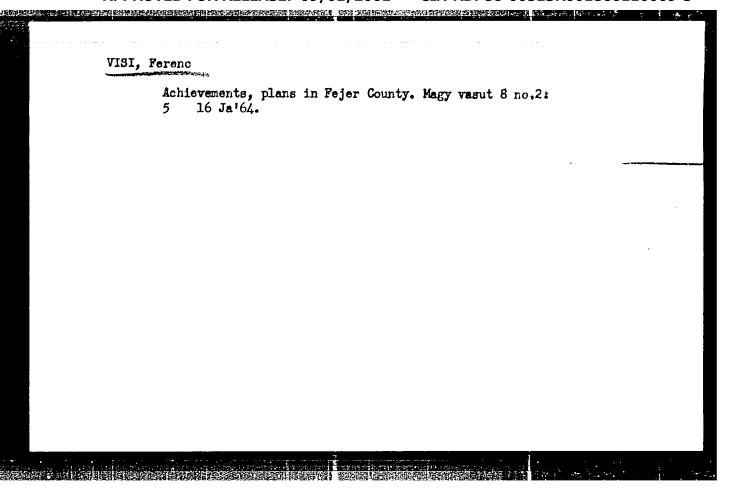


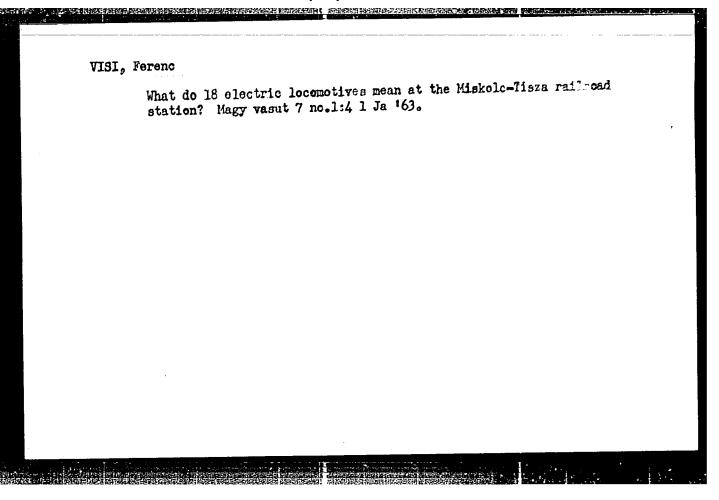
VISI, Ferenc; RODONYI, Karoly

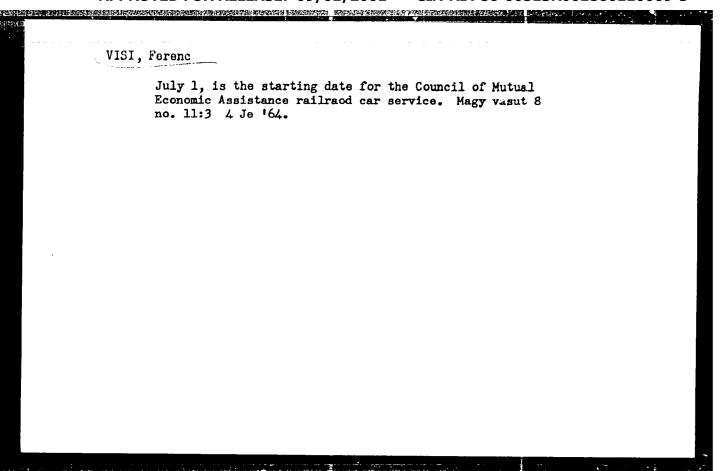
The establishment of the joint freight car service has been advantageous to all the Council of Mutual Economic Assistance countries. Magy vasut 8 no.2: 1 16 Ja 64.

1. MAV vezerigazgato (for Rodonyi).





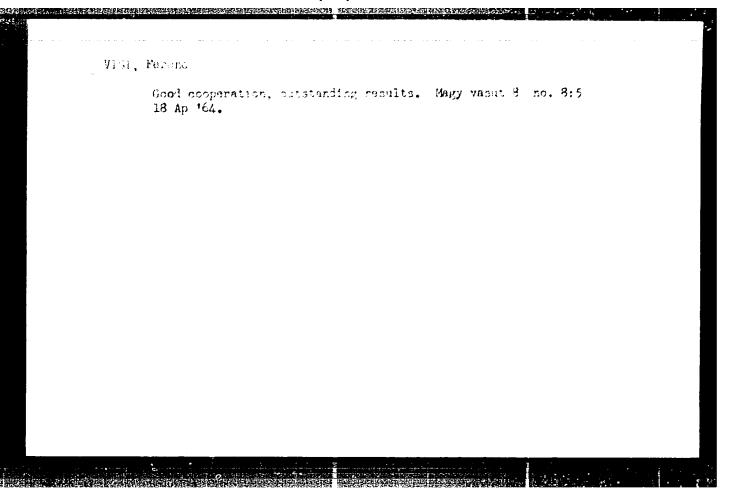


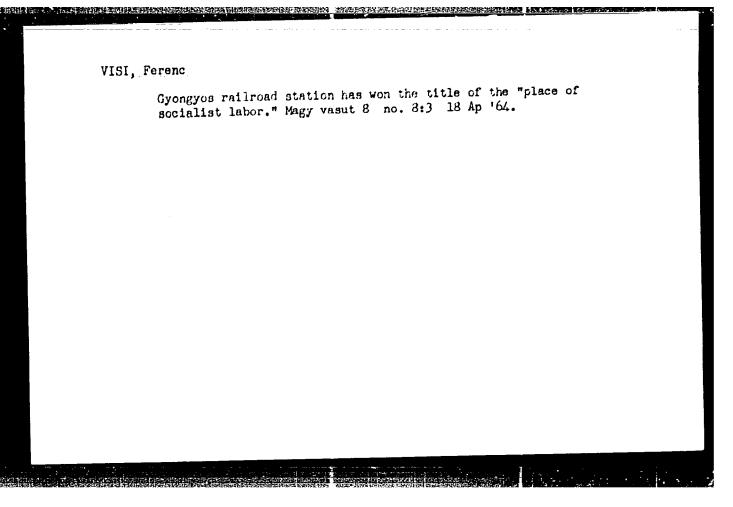


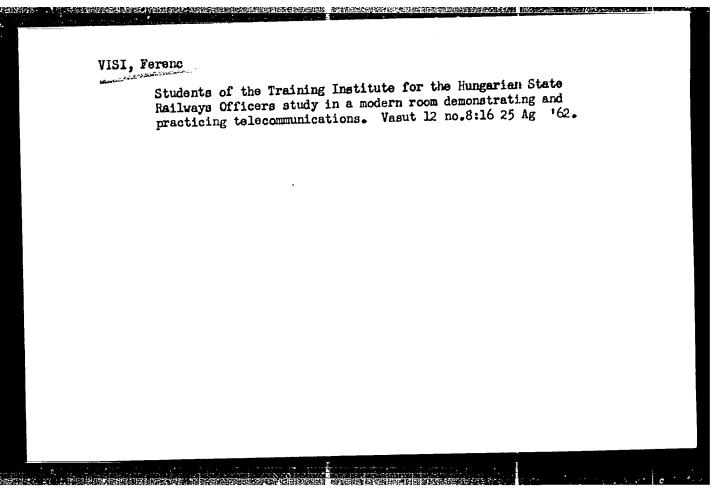
AND SECOND PROPERTY OF THE PRO

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Wage adjustment and better work. Ibid.:6







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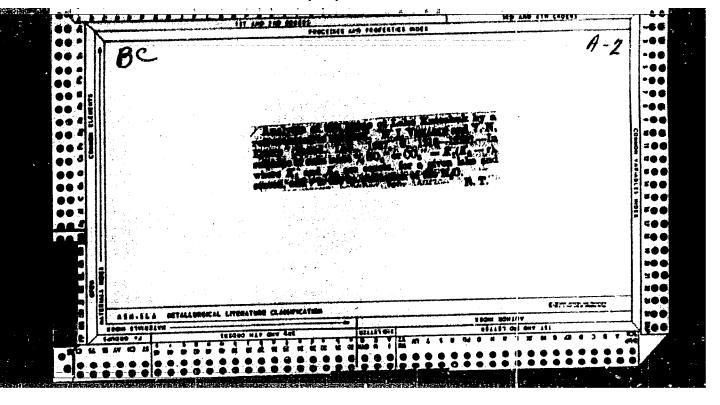
VISI, T.

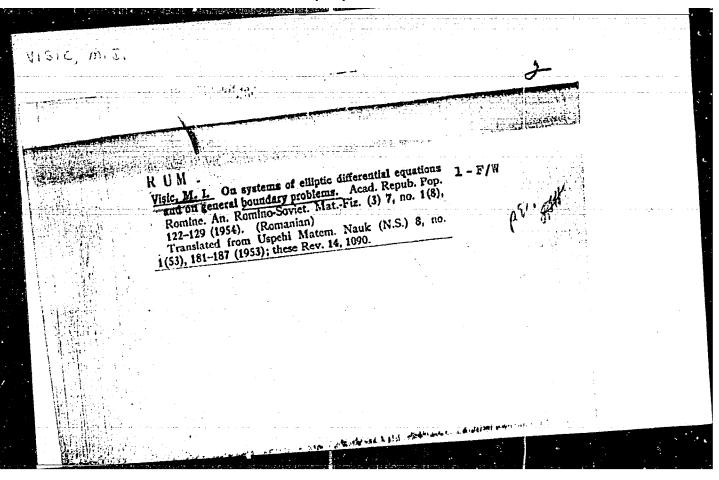
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AUTHOR:

Višić, Stanislav

TITLE:

Mineralization and uranium deposits in the vicinity

of the Bukulje Mountains

PERIODICAL: Referativnyy zhurnal, Geofizika, no. 9, 1962, 45, abstract 9A297 (Radovi Sek. istraž nukl. i drugih mineral. sirovina, 1, no. 1, 1961, 45-66 (Serbo-Croat.;

summary in Fr.))

TEXT: The geologic structure and the results of radiometric investigations of a granite massif in the Bukulje Mountains are described. / Abstracter's note: Complete translation. 7

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DAN SSSR, Vol 92, No 2, pp 205-208

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Cyclopropanes and cyclobutanes. Part 16: \(\rho \) -chloro- and

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\[\rho \) -bromophenylcyclobutanes. Zhur. ob. khim. 31 no. 2:378-382

\[\rho \) -bromophenylcyclobutanes. Zhur. ob. khim. 31 no. 2:378-382

\[\rho \) -bromophenylcyclobutanes. (MIRA 14:2)

\[\rho \) 1. Moskovskiy gosudarstvennyy universitet.

\[\ldot \) (Cyclopropane) (Cyolobutane)

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nauk; PLAKS, A.V., kand. tekhn. nauk, dots.; PRIVALOV, V.V.,
kand. tekhn. nauk; TREYMUNDT, N.D., kand. tekhn. nauk; VISIN,
N.G., kand. tekhn.nauk, retsenzent; KUCHMA, K.G., kand. tekhn.
nauk, retsenzent; FAMINSKIY, G.V., kand. tekhn.nauk, retsenzent;
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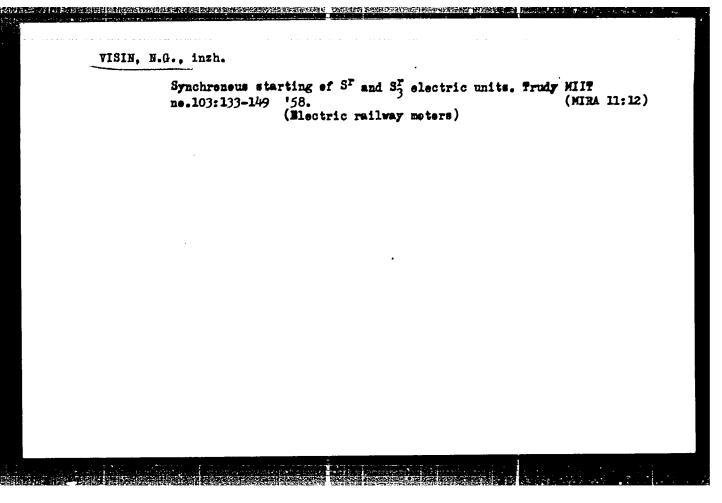
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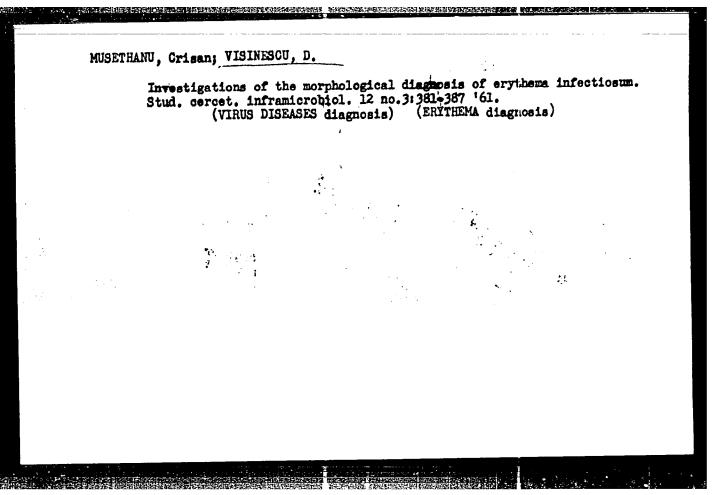
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Abstract [Authors English summary modified]: The respiratory viral epidemic of February-March 1962 was found to be of mixed character. In some patients the influenza virus was isolated and in others a large pararickettsial-like virus. The authors consider the latter to be the causal agent of the primal infection, which may give rise to foci that will produce later cardiovascular accidents and may be accompanied by severe neurological lesions. It is pointed out that the infection should be treated when it first attacks and before the foci are established.

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